of ether and stirred 10 minutes. The MnO₂ was prepared by the method of Henbest et al. (1957). The mixture was centrifuged and the liquid decanted. Pentane (40 ml.) was added to the solid MnO₂, stirred, and decanted. The supernatants were combined, filtered, dried over Na₂SO₄, and concentrated to a crude product (weight 0.242 gram). GLC analysis showed the crude product was 80% pure. The infrared and MS data were consistent with those of nona-trans-2, cis-6-dienal which the authors isolated from cucumbers. The MS data were consistent with those furnished by Forss (1967). The infrared spectra were consistent with published data (Forss et al., 1962a; Jutz, 1959).

The MS of the purified compound were run on a Consolidated 21-620 mass spectrometer. The molecular ion was 138 (low intensity), the base peak was 41, and the major jons in order of intensity were 70, 69, 81, 79, 94, 109, and 95. The infrared data were obtained on a Perkin-Elmer 237 double-beam grating instrument. The infrared spectra are reported in microns with the size of the maxima abbreviated as S meaning strong, M medium, and W weak. The infrared spectra found for the purified compound in the 5- to 15-micron range were: S (5.9), M (6.11, 6.88, 8.51, 8.85, 9.05, 10.25), W (6.78, 7.05, 7.15, 7.3, 7.52, 7.69, 7.94, 9.35, 9.90, 11.43, 12.1, 12.48, 13.89).

LITERATURE CITED

- Forss, D. A., International Flavors and Fragrances, New York, private communication, 1967.
- Forss, D. A., Dunstone, E. A., Horwood, J. F., Stark, W., Aust. J. Chem. 15, 163 (1962a).
- Forss, D. A., Dunstone, E. A., Ramshaw, E. H., Stark, W., J. Food Sci. 27, 90 (1962b).
- Forss, D. A., Hancox, N. C., *Aust. J. Chem.* 9, 420 (1957). Henbest, H. B., Jones, E. R. H., Owen, T. C., *J. Chem. Soc.*
- 1957, p. 4909.
- Jutz, C., Chem. Ber. 92, 1983 (1959).
- Nobuhara, A., Matsui, M., Agr. Biol. Chem. 30 (11), 1087 (1966).
- Sondheimer, F., J. Am. Chem. Soc., 74, 4040 (1952).
- Takei, S., Ono. M., J. Agr. Chem. Soc. Japan 15, 193 (1939); C.A. 33, 6524⁷ (1939).
- Takei, S., Ono, M., Kuroiwa, Y., Takahata, T., Sima, T., J. Agr. Chem. Soc. Japan 14, 717 (1938); C.A. 33, 2558³ (1939).

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Received for review February 19, 1968. Accepted April 25, 1968. Reference to a company or product name does not imply approval or recommendation of the product by the U. S. Department of Agriculture to the exclusion of others that may be suitable.

Correction

SOME CHEMICAL PROPERTIES OF MUNIDA GRE-GARIA AND EUPHAUSIA SUPERBA

In this article by P. R. Burkholder et al. [J. AGR. FOOD CHEM. 15, 718 (1967)], because of some errors in decimal points in Table I, the following corrected Table I is published.

Table I. Proximate Composition of Munida gregaria and Euphausia superba

	Results ^a	
	Munida	Euphausia
	%	
Ash	17.61	13.13
Crude protein ^b	(14.13)	(63.87)
Protein (amino N \times 6.25)	13.19	37.75
Fat (ether extract)	38.08	17.79
Fiber (chitin) ^c	19.77	12.31
Carbohydrate	11.34	19.02
	Per 100 G.	
Caloric value ^d	441	387
" Calculated to moisture-free basis.		

^b Values in parentheses are based on the untenable common assumption that Kjeldahl N \times 6.25 = protein. ^c Lovell *et al.*, J. AGR. FOOD CHEM. **16**, 204 (1968). ^d Based on caloric equivalents of 9, 4, and 4 per gram for fat, protein, and carbohydrate, respectively. Chitin is assumed to be

noncalorigenic.

Also, in the text, page 718, second column, line 27 should read 11.34 and 19.02% rather than 44.98 and 57.92%, respectively, and page 719, first column, line 2 should read 17.79% rather than 1.77%.